

5,6-Diphenyl-3-(2-pyridyl)-1,2,4-triazine

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In the title compound, C₂₀H₁₄N₄, the triazine ring adopts a slight twist conformation. In the crystal structure, intermolecular C—H···N interactions connect the molecules into chains along the *b* axis. The crystal packing is also stabilized by π – π and C—H··· π interactions.

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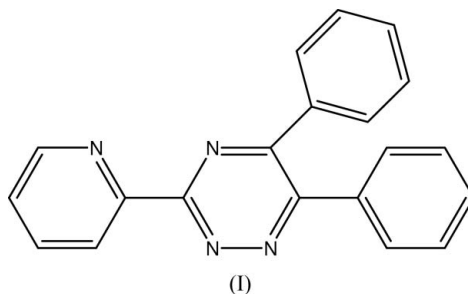
Comment

3,5,6-Trisubstituted-1,2,4-triazines, such as the title compound, (I), represent a principal class of *N*-donor heterocyclic ligands that exhibit interesting pharmacological properties such as blood platelet aggregation inhibition, antiviral, anti-HIV and anticancer (leukemia and ovarian) activity (Soudi *et al.*, 2005; Mashaly *et al.*, 1999). 1,2,4-Triazines have also been used in analytical chemistry to determine the concentration of some trace metal ions (Almog *et al.*, 1996; Croot & Hunter, 2000).

Key indicators

Single-crystal X-ray study
T = 100 K
 Mean σ (C—C) = 0.001 Å
R factor = 0.047
wR factor = 0.131
 Data-to-parameter ratio = 24.8

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.



Bond lengths and angles in (I) (Fig. 1) have normal values (Allen *et al.*, 1987). The C15–C20 and C9–C14 phenyl rings are attached to the triazine ring at atoms C3 and C2, with torsion angles N3–C3–C15–C20 = 51.09 (12)° and C3–C2–C9–C14 = 33.36 (14)°. The triazine ring adopts a slight twist conformation, with puckering parameters (Cremer & Pople, 1975) *Q* = 0.097 (1) Å, θ = 92.2 (6) and φ_2 = 333.1 (5)°.

In the crystal structure, the molecules are linked by intermolecular C—H···N intermolecular interactions, forming chains along the *b* axis (Table 1 and Fig. 2). The crystal packing is stabilized by C—H··· π interactions (Table 1) involving the triazine ring (centroid *Cg* 1) and the C4/N4/C5–C8 (centroid *Cg* 2), C9–C14 (centroid *Cg* 3) and C15–C20 (centroid *Cg* 4) phenyl rings. In addition, the crystal structure is further stabilized by π – π interactions involving the triazine ring at (*x*, *y*, *z*) and C4/N4/C5–C8 at (–*x*, 2 – *y*, –*z*), with a centroid–centroid distance of 3.5434 (6) Å.

Experimental

Recently, we reported the crystal structure of the Mn^{II} and Cd^{II} complexes of 5,6-diphenyl-3-(2-pyridyl)-1,2,4-triazine (Eltayer, Teoh

& Yamin, 2006; Eltayeb, Teoh, Teh *et al.*, 2006) and planned to prepare the Zn^{II} complex by stirring 5,6-diphenyl-3-(2-*pridyl*)-1,2,4-triazine (0.310 g, 1.0 mmol), as purchased from Acros, and ZnCl₂ (0.68 g, 0.5 mmol) together at room temperature for 1 h in a mixture of 20 ml of CH₂Cl₂ and 10 ml of THF. The resulting yellow solution was filtered, and yellow crystals formed after a few days of slow evaporation of the solvent at room temperature. Unfortunately, the crystals were of the starting material and not the Zn^{II} complex.

Crystal data

C₂₀H₁₄N₄ Z = 4
 M_r = 310.35 D_x = 1.352 Mg m⁻³
 Monoclinic, P2₁/c Mo Kα radiation
 a = 11.5830 (3) Å μ = 0.08 mm⁻¹
 b = 11.0381 (2) Å T = 100.0 (1) K
 c = 11.9550 (3) Å Block, yellow
 β = 94.127 (1)° 0.64 × 0.21 × 0.20 mm
 V = 1524.53 (6) Å³

Data collection

Bruker SMART APEX2 CCD area-detector diffractometer 38592 measured reflections
 5386 independent reflections
 ω scans 4472 reflections with I > 2σ(I)
 Absorption correction: multi-scan R_{int} = 0.039
 (SADABS; Bruker, 2005) θ_{max} = 32.3°
 T_{min} = 0.887, T_{max} = 0.984

Refinement

Refinement on F² w = 1/[σ²(F_o²) + (0.0737P)² + 0.4168P]
 R[F² > 2σ(F²)] = 0.047 where P = (F_o² + 2F_c²)/3
 wR(F²) = 0.131 (Δ/σ)_{max} < 0.001
 S = 1.03 Δρ_{max} = 0.50 e Å⁻³
 5386 reflections Δρ_{min} = -0.22 e Å⁻³
 217 parameters
 H-atom parameters constrained

Table 1

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
C7—H7A...N4 ⁱ	0.93	2.52	3.322 (1)	145
C6—H6A...Cg4 ⁱ	0.93	2.82	3.470 (1)	128
C13—H13A...Cg1 ⁱⁱ	0.93	3.00	3.487 (1)	115
C14—H14A...Cg1 ⁱⁱ	0.93	3.32	3.661 (1)	104
C17—H17A...Cg3 ⁱⁱⁱ	0.93	2.65	3.478 (1)	149
C20—H20A...Cg2 ⁱⁱ	0.93	3.10	3.685 (1)	122

Symmetry codes: (i) -x, y + 1/2, -z + 1/2; (ii) x, -y + 3/2, z - 1/2; (iii) -x + 1, -y + 1, -z. Cg1, Cg2, Cg3 and Cg4 are the centroids of the triazine, C4/N4/C5–C8, C9–C14 and C15–C20 rings, respectively.

H atoms were positioned geometrically and treated as riding, with C—H = 0.93 Å and U_{iso}(H) = 1.2U_{eq}(C).

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2003).

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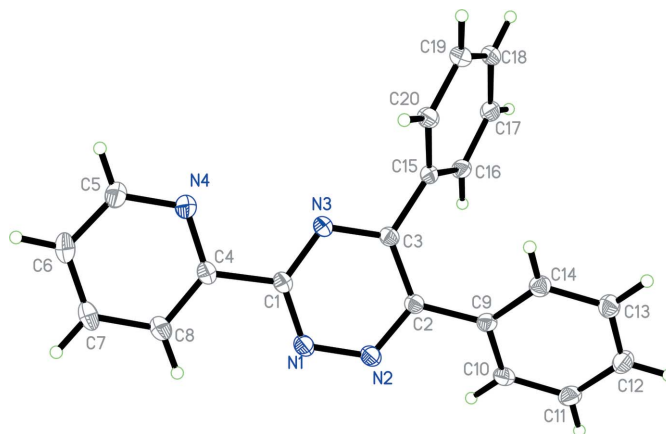


Figure 1 The molecular structure of (I), showing 50% probability displacement ellipsoids and the atomic numbering.

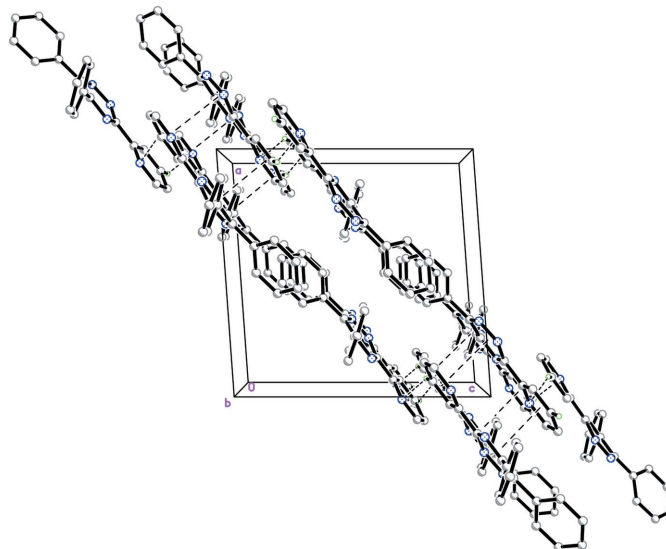


Figure 2 The crystal packing of (I), with intermolecular C—H...N hydrogen bonds and π-π interactions shown as dashed lines.

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Almog, J., Hirshfeld, A., Glattstein, B., Sterling, J. & Goren, Z. (1996). *Anal. Chim. Acta*, **322**, 203–208.
 Bruker (2005). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
 Croot, P. L. & Hunter, K. A. (2000). *Anal. Chim. Acta*, **406**, 289–302.
 Eltayeb, N. E., Teoh, S. G., Teh, J. B.-J., Fun, H.-K., Yamin, B. M. & Ibrahim, K. (2006). *Acta Cryst. E* **62**, m3336–m3337.
 Eltayeb, N. E., Guan, T. S. & Yamin, B. M. (2006). *Acta Cryst. E* **62**, m2284–m2286.
 Mashaly, M., Bayoumi, H. A. & Taha, A. (1999). *J. Serb. Chem. Soc.* **64**, 621–635.
 Sheldrick, G. M. (1998). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
 Soudi, A. A., Marandi, F., Morsali, A., Kempe, R. & Hertle, I. (2005). *J. Coord. Chem.* **58**, 1631–1637.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.